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Synthesis and Electroluminescence Properties of Heterocyclic Conjugated Copolymers and Blends

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Improved electroluminescent efficiencies in polymer light-emitting diodes(LEDs) have been reported to be obtained either by constructing multi-layer devices or by using polymer blends in a light emitting layer. In this work, conjugated copolymers, poly((PZV-PV)-co-(CZV-PV)) containing (N-2-ethylhexyl)-3,6-phenothiazinylene vinylene(PZV) unit and (N-2-ethylhexyl)-3,6-carbazolylene vinylene(CZV) unit in different ratio were synthesized by Horner-Emmons reaction utilizing potassium *t*-butoxide. Homopolymers, poly(PZV-PV) and poly(CZV-PV) were also synthesized. Polymer LED's fabricated with poly(PZV-PV) and poly(CZV-PV) homopolymers as light emitting layers exhibited EL maximum at 570nm amd 500nm, respectively. LED's made with poly((PZV-PV)-co-(CZV-PV)) copolymers, however, showed EL maxima shifted to longer wavelengths.

<u>Keywords</u> light-emitting diode; heterocyclic conjugated copolymer; Horner-Emmons condensation; light emitting layers

INTRODUCTION

Conjugated polymers have received considerable attention due to their application in light-emitting diode(LED)s, since R. H. Friend et al. reported LED made from poly(p-phenylene vinylene) (PPV) [1]. The microstructures of conjugated polymers are known to be one of most important factors affecting the emission color, quantum efficiency, and

luminance/voltage profile of LED. In this work, we describe synthesis and electro-optical properties of conjugated copolymers containing phenothiazine and carbazole moiety in the main chain in order to increase the hole injection and transporting property. We also examined the usefulness of Horner-Emmons reaction for obtaining conjugated copolymers with different microstructure for application to the electroluminescence devices.

EXPERIMENTAL

Heterocyclic copolymers, poly((PZV-PV)-co-(CZV-PV)) containing N-(2-ethylhexyl)-3,6-phenothiazinylene vinylene(PZV) unit and N-(2ethylhexyl)-3,6-carbazolylene vinylene(CZV) unit in different ratio were synthesized by Horner-Emmons reaction utilizing potassium tbutoxide [2]. The synthetic scheme of heterocyclic copolymer is shown in FIGURE 1. N-(2-ethylhexyl) phenothiazine(EHPZ) was obtained from phenothiazine and 2-ethylhexyl bromide in dimethylsulfoxide solvent with sodium hydroxide. EHPZ was reacted with POCl₃ and DMF in dichloroethane solvent to give aromatic dialdehyde monomer, N-(2-ethylhexyl)-3,6-diformylphenothiazine(DFPZ). N-(2-ethylhexyl)-3,6-diformylcarbazole(DFCZ) was synthesized by a synthetic method similar of DFPZ that starting from Bis(diethoxyphosphinylmethyl)benzene(PHBZ) was synthesized by reacting α,α -dichloro-p-xylene with triethylphosphite.

The molecular weights of synthesized copolymers were measured by Waters gel permeation chromatograph(GPC). Electroluminescence(EL) spectra and color coordinates of LED's were measured with Spectroscan PR 704 (Photoresearch Inc.). The EL intensity was measured by using Minolta luminance meter (LS-100) at room temperature.

RESULTS AND DISCUSSION

Synthesis and Characterization of Conjugated Polymers

Molecular weight and copolymer composition data of conjugated copolymers synthesized by Horner-Emmons reaction are shown in TABLE 1.

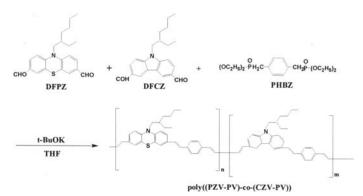


FIGURE 1. Synthetic route to heterocyclic conjugated copolymers.

TABLE 1. Composition and M_W of EL polymers

	''' 1 /		
Conjugated	Feed ratio	Copolymer	MW
Polymers	(DFPZ:DFCZ:PHBZ)	Composition	(g/mol)
Poly(PZV-PV)	50:0:50	50:0:50	15,900
Poly(CZV-PV)	0:50:50	0:50:50	12,100
Poly(PZV-co-CZV)19	5:45:50	9:41:50	15,900
Poly(PZV-co-CZV)37	15:35:50	18:32:50	13,900
Poly(PZV-co-CZV)55	25:25:50	29:21:50	16,900
Poly(PZV-co-CZV)73	35:15:50	38:12:50	17,600
Poly(PZV-co-CZV)91	45:5:50	43:7:50	43,400

The molecular weights of copolymers were in the range of 12,000-43,000 g/mol as determined by GPC with THF as eluent. The low Mw copolymers may be due to the deviation from the stoichiometric balance of two comonomers which is a critical factor controlling the Mw of polymer in the condensation polymerization. The composition of copolymers determined by the elemental analysis was close to the feed ratio of monomers as normally observed in the polycondensation reaction. The good solubility in common organic solvents at room temperature was one of the important properties of the synthesized copolymers. The high solubility of the conjugated copolymers seemed to be due to the presence of heterocyclic aromatic ring in the main chain and also due to the ethylhexyl group in the phenothiazine and carbazole moiety.

Electro-optical Properties of Conjugated Polymers

Polymer LED's fabricated with poly(PZV-PV) and poly(CZV-PV) homopolymers as light emitting layers exhibited EL maximum at 570 nm and 500 nm, respectively. The LED's made with blends of homopolymers as light emitting layers exhibited EL maxima between the those of homopolymers. LED's made with poly((PZV-PV)-co-(CZV-PV)) copolymers, however, showed EL maxima shifted to longer wavelengths. The detailed data on the $\lambda_{\text{max,EL}}$ for the EL polymers are listed in TABLE 2. This may be due to the enhanced exciplex formation in the case of conjugated copolymers with different microstructures of the repeating units in the polymer chain.

TABLE 2. EL properties of conjugated copolymers

	s of conjugated coporymers		
Conjugated	Feed ratio) (nm)	
Polymers	(DFPZ:DFCZ:PHBZ)	$\lambda_{\max, EL}(nm)$	
Poly(PZV-PV)	50:0:50	570	
Poly(CZV-PV)	0:50:50	500	
Poly(PZV-co-CZV)19	5:45:50	578	
Poly(PZV-co-CZV)37	15:35:50	586	
Poly(PZV-co-CZV)55	25:25:50	598	
Poly(PZV-co-CZV)73	35:15:50	592	
Poly(PZV-co-CZV)91	45:5:50	584	
Polymer blend	Poly(PZV-PV):poly(CZV-PV)	$\lambda_{\text{max},\text{EL}}$	
Blend 19	lend 19 1:9		
Blend 55	5:5	564	
Blend 91	9:1	562	

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